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para-Quinone Methide: a New Player in Asymmetric Catalysis

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para-Quinone methides (*p*-QMs)^[1] have been known as reaction intermediates for more than a century.^[2] They consist of a cyclohexadiene moiety in *para*-conjugation with a carbonyl group and an *exo*-methylene component (**C**, Figure 1). *p*-QMs are neutral molecules with an aromatic zwitterionic resonance structure which makes them more reactive than structurally related *para*-quinone **A** and *para*-quinone dimethide **B** (Figure 1). This enhanced electrophilicity^[3] has been used in a variety of medicinal and biological processes such as DNA alkylation, cross-linking and enzyme inhibition.^[4] The *para*-quinone methide moiety is also present in a variety of biologically active compounds and has been proposed as an intermediate in the biosynthesis of a number of natural products.^[5]

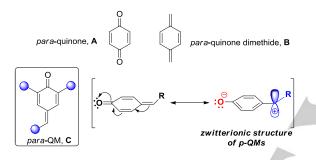
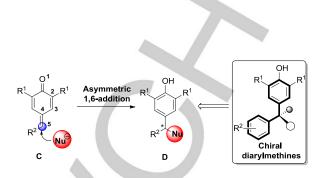


Figure 1 p-Quinone, p-quinone dimethide and p-quinone methide derivatives.

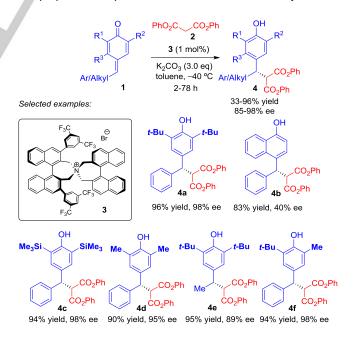
Surprisingly, while *ortho*-quinone methides have been broadly used in asymmetric synthesis,^{[7],[8]} *p*-QMs have received relatively little attention, particularly among catalyzed methods. Until recently, the only enantioselective reactions reported were in the context of asymmetric polymerizations, and low stereoselectivities were observed.^[8] While unsubstituted *p*-QMs are too reactive to be studied in asymmetric catalysis, the use of more stable 2,6-disubstituted *p*-QMs **C** offers the possibility of addressing this challenge. Intermediates **C** represent attractive targets, because asymmetric 1,6-additions^[9] could rapidly afford important chiral diarylmethines (Scheme 1).^[10] This idea has been nicely developed by the groups of Fan and Jørgensen, opening a new entry to enantiomerically enriched benzyl derivatives.

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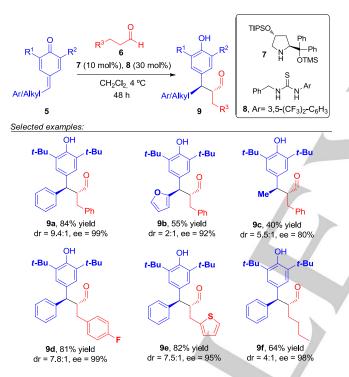
Scheme 1. p-Quinone methides as synthons for chiral diarylmethines.

Fan et al.[11] recently reported the use of 2,6-disubstituted p-QMs 1 as stable prochiral starting materials in a catalytic asymmetric transformation. They reported the 1,6-addition of diphenyl malonate 2 to p-QMs 1 using chiral ammonium phase transfer catalysis.^[12] The use of center-chiral catalysts such as N-bridged cinchona or tartrate-derived ammonium salts was not effective. However, the binaphthyl-modified ammonium bromide with axial chirality, gave excellent results. Several 3. enantiomerically enriched malonates were prepared in good yields and high enantiomeric excesses (Scheme 2). The catalytic system worked well with p-QMs bearing both aromatic and aliphatic substituents. Interestingly, a p-QM derived from the 1,4-naphthoquinone (4b precursor) was described for the first time although only a moderate enantiomeric excess was obtained (4b, 40% ee). A plausible transition-state model was also proposed to explain the observed enantioselectivity.



Scheme 2. Asymmetric 1,6-addition of malonates to p-QMs

Very recently, Jørgensen et al.[13] published a novel approach for the α -alkylation of aldehydes using p-QMs as alkylating agents under organocatalytic enamine catalysis^[14] (Scheme 3). Although the Jørgensen-Hayashi catalyst afforded the products with good yields and high enantioselectivities, the diastereomeric ratios were poor. A new secondary amine 7 was then designed to solve this problem (Scheme 3). The bulky tiisopropylsilyloxy group at C₄, anti to the C₁ substituent, was key to determine the approach of the p-QM to the enamine and therefore to control the diastereoselectivity Additionally, the use of 30 mol% of thiourea 8 significantly improved yields, probably by activation of the carbonyl group on 5 though hydrogen bonding. A wide range of α-alkylated aldehydes 9 were prepared in good yields (40-90%), moderate to good diastereoselectivities (dr = 2:1-11:1) and excellent enantiocontrol (ee = 80-99%). Despite the high reactivity of p-QMs 5, the process was highly chemoselective and N-alkylation of catalyst 7 was not observed.

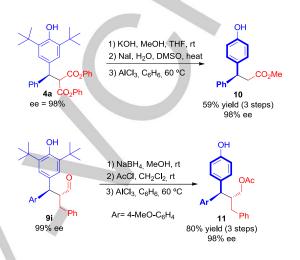


Scheme 3. Asymmetric enamine addition to *p*-QMs. TIPS= triisopropylsilyl. TMS= trimethylsilyl.

In most cases, both Fan and Jørgensen used 2,6-di-*tert*butyl *para*-quinone methides^[4] due to their inherent stability and facile preparation. Notably, the role of the *tert*-butyl group is doubly important because it stabilizes the starting *p*-QMs, making their isolation far more practical, and can be easily removed afterwards with AlCl₃ (Scheme 4).

In summary, *p*-QMs have been successfully used for the first time in asymmetric catalysis in two different processes: the 1,6-addition of phenyl malonate under phase transfer catalysis, and the α -alkylation of aldehydes using enamine catalysis. In both cases, the products were obtained in good yields and high

enantioselectivities. These two publications represent a powerful entry towards the synthesis of enantiomerically enriched diarylmethines and will undoubtedly inspire the development of other catalytic asymmetric transformations in the future. Applications of *p*-quinone methides with other modes of activation in organocatalysis as well as metal-catalyzed reactions are still waiting to be explored.



Scheme 4. Elimination of the t-Butyl group on the phenolic ring.

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Keywords: quinone methide • asymmetric catalysis• reaction intermediate • organocatalysis • 1,6-addition

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HIGHLIGHT

p-QMs have been successfully used in asymmetric organocatalysis. Particularly, the asymmetric 1,6addition of phenyl malonate and different aldehydes 2,6to disubstituted p-QMs has provided a rapid access to important chiral diarylmethines, highlighting the these importance of synthetic intermediates. These new structures will open up the development of important asymmetric transformations in the future.

