

Repositorio Institucional de la Universidad Autónoma de Madrid

https://repositorio.uam.es

Esta es la **versión de autor** del artículo publicado en: This is an **author produced version** of a paper published in:

Chemical Communication 54.22 (2018): 2781-2784

DOI: https://doi.org/10.1039/C8CC00759D

Copyright: © 2018 The Royal Society of Chemistry

El acceso a la versión del editor puede requerir la suscripción del recurso

Access to the published version may require subscription

Asymmetric Vinylogous Mukaiyama Aldol Reaction of Isatins under Bifunctional Organocatalysis: Enantioselective Synthesis of Substituted 3-Hydroxy-2-Oxindoles

Víctor Laina-Martín,^{a4} Jorge Humbrías-Martín,^{a4} Jose A. Fernández-Salas,^{a*} and José Alemán^{a,b*}

A highly enantioselective organocatalytic vinylogous Mukaiyama aldol reaction of silyloxy dienes and isatins under bifunctional organocatalysis is presented. Substituted 3-hydroxy-2-oxindoles are synthesised in good yields and enantioselectivities. These synthetic intermediates are used for the construction of more complex molecules with biological properties such as the formal synthesis of a CB2 agonist presented.

The development of efficient and practical strategies for the stereoselective construction of privileged structures is an ongoing objective and still holds a preferred position in organic chemistry research.¹ Among these privileged structures, 3-substituted-3-hydroxyindolin-2-ones have attracted much attention from organic and medicinal These heterocyclic motifs constitute the core chemists. structure of a family of diverse natural and non-natural products² with biological and pharmaceutical activities.³ Therefore, the asymmetric synthesis of 3-hydroxy-2-oxindole derivatives such as Donaxaridine⁴ or Convolutamyde⁵ has been widely pursued (C, Scheme 1). The substitution at the C3 position as well as the configuration of the quaternary center are known to be crucial for the activity of this family of compounds.^{2c} Consequently, the functionalisation of the isatin (C3) at that position is of great interest to the research community and is currently an open area in asymmetric catalysis. In order to face this challenge, the direct addition of different nucleophiles to isatins under catalytic conditions has been the method of choice for different research groups,⁶ including enantioselective direct aldol reactions.⁷ The

^b Institute for Advanced Research in Chemical Sciences (IAdChem), Universidad Autónoma de Madrid, 28049 Madrid, Spain.

 $^{\boldsymbol{\varkappa}}$ These authors contributed equally to this work.

‡ Electronic Supplementary Information (ESI) available: See DOI: 10.1039/ x0xx00000x.



Scheme 1. Previous works (A), present work (B) and importance of the isatin core in biologically active molecules and commercial drugs (C).

construction of these challenging quaternary centers has also been attempted via the Mukaiyama-aldol reaction under either metal or organo-catalytic conditions (A, Scheme 1).⁸ In recent years, the organocatalysed asymmetric vinylogous aldol reaction⁹ has appeared as a powerful tool for the generation of multifunctional chiral alcohols with easily derivatizable unsaturated carbon chains that allow the construction of more complex structures (B, Scheme 1).¹⁰ The organocatalysed vinylogous Mukaiyama-aldol (VMA) reaction has been widely developed and has become the preferred method for the stereoselective synthesis of aldol type products.¹¹ This approach is the favoured protocol to furnish vinylogous aldol adducts, overcoming regioselectivity issues and multistep synthesis. Despite the importance of these aldolic products (C, Scheme 1), no catalytic enantioselective version of the VMA reaction to isatins has been described. To the best of our knowledge, only Kobayashi et al.^{5a} have described the enantioselective synthesis of Covolutamydes B and E, using silvloxy diene derivatives bearing a chiral inductor. Therefore, we believe that the development of an efficient asymmetric

 ^a Departamento de Química Orgánica (Módulo 1), Facultad de Ciencias, Universidad Autónoma de Madrid, 28049-Madrid, Spain. E-mail: jose.aleman@uam.es; webpage:
www.uam.es/jose.aleman

organocatalytic VMA reaction to isatins would be highly desirable. The adducts obtained which can bear larger unsaturated alkyl chains will facilitate easy functionalisations, making this approach a very attractive method. Therefore, this approach would have the potential to efficiently reach interesting platforms for the synthesis of more complex molecules with proven properties.

In this paper we describe the enantioselective addition of silyl dienol ethers to isatins under bifunctional organocatalytic conditions (B, Scheme 1). This VMA reaction provides direct and rapid access to δ -hydroxylated α,β -unsaturated carbonyls, which are themselves prevalent arrays in biologically relevant natural molecules, or are privileged platforms for their construction.

Based on our experience in bifunctional catalysis,¹² we began our investigations by studying the reaction of isatins (**1a-1d**) with a trimethyl silyl enolate derivative **2a** and different thiourea and squaramide bifunctional organocatalysts, in

Table 1. Screening reaction conditions for the addition of 2a to 1a-d in the presence of different catalysts 3.°



^a All the reactions were performed in 0.1 mmol scale in presence of H₂O (X equiv.) in 0.3 mL of solvent. ^b Determined by ¹H NMR analysis of the crude. ^c Determined by SFC chromatography. ^d Experiment carried out with 20 mol % of **3a**. ^e Experiment carried out with 5 mol % of **3a**. presence of H_2O (Table 1).¹³ Firstly, we examined the reaction of the *N*-unprotected isatin **1a** in dichloromethane, and observed moderate stereocontrol but full regioselectivity. Only 1,5 product **4a** was observed in the crude mixture without detection of the corresponding 1,3 product.

We then decided to investigate the impact of the Nsubstitution of the isatin. Therefore, we carried out the reaction of methyl and benzyl N-protected isatin derivatives (entries 2 and 3). In both cases, methyl (1b) and benzyl (1c), showed an increase in enantioselectivity. The benzyl-protected isatin showed the best enantiomeric ratio, even when the catalyst loading was decreased by up to 10 mol % (entry 4). We then examined different bifunctional organocatalysts (entries 4-9). The squaramide-based catalyst (Rawal's catalyst, 3b) showed low stereocontrol, while the thiourea-catalysts (3a, 3c-f) presented better enantiomeric ratios. Among them, the cinchone organocatalysts led to the highest *e.r.* (entries 6-9). Different solvents were then studied (entries 10-12 and see ESI). THF was found to be the optimum solvent (entry 12) with a 91:9 enantiomeric ratio. The role of water is crucial for this reaction.¹² In the absence of water almost no conversion occurred, whereas the presence of 1 equiv. led to 4c with a 57 % conversion (entries 13 and 14). However, an excess of water (6 equiv.) produced the hydrolysis of 2a and provoked a decrease in the final conversion of 4c (entry 15). A lower temperature (-30 °C) slightly enhanced the e.r. (entry 13-16). Interestingly, the bulkier N-p^tBu-benzyl group yielded the desired product 4d, with complete enantiocontrol (entry 17), while a lower catalyst loading (5 mol%) strongly decreased the final conversion (entry 18).

Once the reaction conditions had been optimised (entry 17, Table 1), we then studied the scope of the reaction considering different substitutions in the aromatic ring of the isatin (Table 2). Electron-donating groups, such as the 5-OMe, 5-Me and 6-OMe substituents in the aromatic ring of the isatin, gave 4e, 4f and 4g, with good yields and enantiomeric ratios. The reaction proceeded smoothly with isatins bearing electron-withdrawing groups (5-F, 6-CF₃ and 5-NO₂), and the desired products: 4h, 4m and 4r were obtained in good yields and with high stereocontrol. The protocol enabled access to the corresponding 3-hydroxyindoles bearing more synthetically useful halogens such as bromo groups (4h and 4j) in the aromatic ring. The presence of any substituent at position 7 of the phenyl ring could have been affecting the efficient coordination of the substrate to the catalyst, leading to a slightly lower enantiocontrol (4k, 4n and 4p). Therefore, we envisioned that a smaller protecting group at the nitrogen atom of the isatin could potentially prevent the plausible steric interaction with the very bulky N-protecting group. To our delight, the N-benzyl protected derivatives allowed access to the desired 7-substituted adducts with higher enantiomeric ratios (4l, 4o and 4q). Other silyl nucleophiles were also tested in order to gain a better understanding of the process and to expand the scope of the reaction. The reaction with the silylenolate **2b** led to the ketones **4s** and **4t** in good yields but with





 a All the reactions were performed in 0.1 mmol scale in presence of 3 equiv. of H_2O in 0.3 mL of THF. Enantiomeric ratio was determined by SFC chromatography. b Reaction carried out at room temperature.

lower enantioselectivity. Unfortunately, the result obtained using the silyl enolate formed from the analogous ester, gave **4u** in very low enantiomeric excess. The absolute configuration of 3-hydroxyindole products **4** was assigned by correlation with a known compound in the literature (**4s**) and was determined as 3*R*, assuming the same stereochemical outcome for the rest of compounds **4** (*see* ESI).

This vinylogous Mukaiyama protocol enables access to versatile building blocks using commercially available materials, which can be used for the synthesis of natural and bioactive products. A simple alkylation of the readily available and cheap isatin **5** (8 euros/gram) with the bromocyclopropyl derivative led to the *N*- alkylated isatin (**1s**) in a good yield. Then, the VMA reaction gave the aldehyde **4v** with good yield and high enantioselectivity. The subsequent NaBH₄ reduction of the aldehyde to the alcohol and the Pd(C) catalysed hydrogenation of the double bond led to the aliphatic primary alcohol **6** in an excellent yield without any loss of enantioselectivity. This alcohol **6** is the key intermediate in the synthesis of a potential



drug candidate (CB2 agonist)^{3g,9a} for reducing neuropathic and bone pain.

In summary, we have described the vinylogous Mukaiyama aldol reaction of silvl dienolates to isatins under bifunctional organocatalysis, which is able to direct both coupling partners. The reaction proceeded with good yields and high enantioselectivities with a reasonably high range of differently substituted isatins. The methodology described offers an easy and straightforward method for the synthesis of enantioenriched biologically relevant 3-substituted-2-oxindole Moreover, the derivatives. products obtained have demonstrated that they can be easily transformed to biologically valuable compounds.

The Spanish Government (CTQ2015-64561-R) is acknowledged. V. L.-M. thanks the Autonomous university of Madrid for a predoctoral fellowship (FPI-UAM). J. A. F.-S. thanks the Spanish Goverment for a Juan de la Cierva Contract.

Notes and references

- 1 V. Farina, J.T. Reeves, C. H. Senanayake and J. J. Song, *Chem Rev.* 2006, **106**, 2734-2793.
- For natural products, see: (a) C. Marti and E. M. Carreira, *Eur. J. Org. Chem.* 2003, 2209–2219; (b) C. V. Galliford and K. A. Scheidt, *Angew. Chem. Int. Ed.* 2007, **46**, 8748–8758; (c) S. Peddibhotla, *Curr. Bioact. Compd.* 2009, **5**, 20–38.
- 3 For selected examples of drug candidates, see: (a) R. B. Labroo and L. A. Cohen, J. Org. Chem. 1990, 55, 4901-4904; (b) P. Hewawasam, N. A. Meanwell, V. K. Gribkoff, S. I. Dworetzky and C. G. Boissard, Bioorg. Med. Chem. Lett. 1997, 7, 1255-1260; (c) Y. Koguchi, J. Kohno, M. Nishio, K. Takahashi, T. Okuda, T. Ohnuki and S. Komatsubara, J. Antibiot. 2000, 53, 105–109; (d) Y.-Q. Tang, I. Sattler, R. Thiericke, S. Grabley and X.-Z. Feng, Eur. J. Org. Chem. 2001, 261–267; (e) T. Tokunaga, W. E. Hume, T. Umezome, K. Okazaki, Y. Ueki, K. Kumagai, S. Hourai, J. Nagamine, H. Seki, M. Taiji, H. Noguchi and R. Nagata, J. Med. Chem. 2001, 44, 4641-4649; (f) P. Hewawasam, M. Erway, S. L. Moon, J. Knipe, H. Weiner, C. G. Boissard, D. J. Post- Munson, Q. Gao, S. Huang, V. K. Gribkoff and N. A. Meanwell, J. Med. Chem. 2002, 45, 1487-1499; (g) R. L. Hinman and C. P. Bauman, J. Org. Chem. 1964, 29, 2431-2437; (h) P. J. D. Dollings, A. F. Donnell, A. M.Gilbert, M. Zhang,

B. L. Harrison, C. J. Stanton, S. V. O'Neil, L. M. Havran and D. C. Chong, 2010, WO2010077839.

- 4 H. B. Rasmussen and J. K. MacLeod, J. Nat. Prod. 1997, 60, 1152-1154.
- 5 (a) T. Nakamura, S.-I Shirokawa, S. Hosokawa, A. Nakazaki and S. Kobayashi, Org. Lett. 2006, 8, 677-679; (b) S. Miah, C. J. Moody, I. C. Richards and A. M. Z. Slawin, J. Chem. Soc. Perkin Trans. 1, 1997, 2405– 2412; (c) S. Nakamura, N. Hara, H. Nakashima, K. Kubo, N. Shibata and T. Toru, Chem. Eur. J. 2008, 14, 8079-8081; (d) A. V. Malkov, M. A. Kabeshov,M. Bella, O. Kysilka, D. A. Malyshev, K. Pluháčková and P. Kočovský, Org. Lett. 2007, 9, 5473-5476.
- 6 For a review, see: S. Mohammadi, R. Heiran, R. P. Herrera, and E. Marqués-López, *ChemCatChem*, 2013, **5**, 2131-2148.
- For selected examples, see: (a) M. Raj, N. Veerasamy and V. K. Singh, *Tetrahedron Lett.* 2010, **51**, 2157-2159; (b) U. V. S. Reddy, M. Chennapuram, K. Seki, C. Seki, B. Anusha, E. Kwon, Y. Okuyama, K. Uwai, M. Tokiwa, M. Takeshita and H. Nakano, *Eur. J. Org. Chem.*, 2017, 3874-3885; (c) Q. Guo, M. Bhanushali and C.-G. Zhao, *Angew. Chem. Int. Ed.* 2010, **49**, 9460-9464; (d) J. Guang, Q. Guo and J. C.-G. Zhao, *Org. Lett.* 2012, 14, 3174-3177; H. Lu, J. Bai, J. Xu, T. Yang, X. Lin, J. Li, F. Ren, *Tetrahedron*, 2015, **71**, 2610-2615.
- 8 For organocatalysis, see: (a) N. V. Hanhan, A. H. Sahin, T. W. Chang, J. C. Fettinger and A. K. Franz, *Angew. Chem. Int. Ed.* 2010, **49**, 744-747; (b) Y. –L. Liu, F. –M. Liao, Y. –F. Niu, X.-L. Zhao and J. Zhou, *Org. Chem. Front.* 2014, **1**, 742-747; (c) Y. L. Liua and J. Zhou, *Chem. Commun.* 2012, **48**, 1919-1921; for metal-catalysis, see: (d) K. Aikawa, S. Mimura, Y. Numata, and K. Mikami, *Eur. J. Org. Chem.* 2011, 62-65.
- 9 The asymmetric vinylogous aldol reaction to isatins starting from allylic carbonyl derivatives has been previously reported. See: a) B. Zhu, W. Zhang, R. Lee, Z. Han, W. Yang, D. Tan, K.-W. Huang and Z. Jiang, *Angew. Chem. Int. Ed.* 2013, **52**, 6666-6670. b) C. Cassani and P. Melchiorre, *Org. Lett.*, 2012, **14**, 5590-5593.
- For selected reviews, see: (a) S. E. Denmark, J. R. Heemstra, Jr. and G. L. Beutner, *Angew. Chem., Int. Ed.* 2005, **44**, 4682-4698; (b) G. Casiraghi, L. Battistini, C. Curti, G. Rassu, and F. Zanardi, *Chem. Rev.* 2011, **111**, 3076-3154; (c) Bisai, V. *Synthesis* 2012, **44**, 1453-1463.
- 11 For a review, see: (a) S. V. Pansare and E. K. Paul, *Chem. Eur. J.* 2011, **17**, 8770-8779; (b) M. Kalesse, M. Cordes, G. Symkenberga and H.-H. Lu, *Nat. Prod. Rep.*, 2014, **31**, 563-594; (c) S. Hosokawa, *Tetrahedron Lett.* 2018, **59**, 77-88. For selected examples, see: (d) R. P. Singh, B. M. Foxman and L. Deng, *J. Am. Chem. Soc.* 2010, **132**, 9558-9560; (e) N. Zhu, -b. C. Ma, Y. Zhang, W. Wang, *Adv. Synth. Catal.* 2010, **352**, 1291-1295; (f) V. Bhasker, M. Gravel, V. H. Rawal, *Org. Lett.* 2005, **7**, 5657-5660; (g) S. E. Denmark and G. L. Beutner, *J. Am. Chem. Soc.*, 2003, **125**, 7800-7801; (h) M. Sickert and C. Schneider, *Angew. Chem. Int. Ed.*, 2008, **47**, 3631-3634; (i) L. Ratjen, P. García-García, F. Lay, M. E. Beck and B. List, *Angew. Chem. Int. Ed.*, 2011, **50**, 754-758.
- (a) C. Jarava-Barrera, F. Esteban, C. Navarro-Ranninger, A. Parra and J. Alemán, *Chem. Commun.* 2013, **49**, 2001. (b) A. Parra, R. Alfaro, L. Marzo, A. Moreno-Carrasco, J. L. García Ruano and J. Alemán, *Chem. Commun* 2012, **48**, 9759. (c) V. Marcos, J. Alemán, J. L. García Ruano, F. Marini, T. Marcello *Org. Lett.* 2011, **13**, 3052. (d) M. Frías, R. Mas-Balleste, S. Arias, C. Alvarado and J. Alemán, *J. Am. Chem. Soc.* 2017, **139**, 672-679. (e) F. Esteban, W. Cieślik, E. M. Arpa, A. Guerrero-Corella, S. Díaz-Tendero, J. Perles, J. A. Fernández-Salas, A. Fraile, J. Alemán, J. *ACS Catal* 2018, **8**, 1884–1890. (f) M. Frias, A. C. Carrasco, A. Fraile, J. Alemán, *Chem. Eur. J.* 2018, DOI: 10.1002/chem.201705218.
- 13 H_2O plays an important role to trigger the aldol reaction and in the interaction with the catalysts. See ESI and ref. 12d.

GRAPHICAL ABSTRACT:

